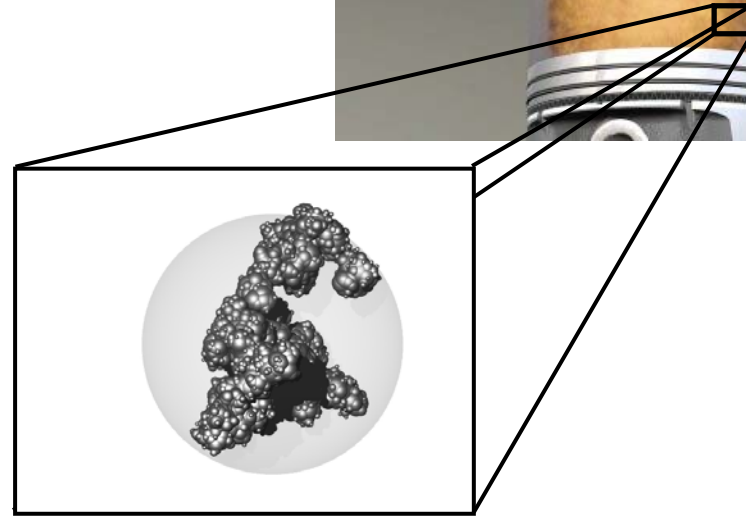
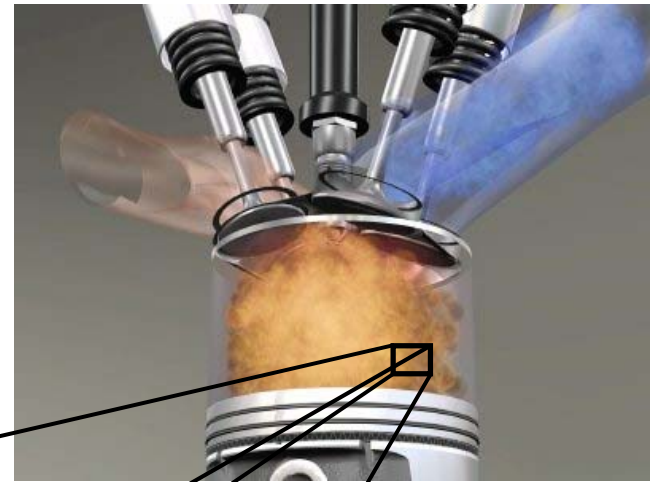


# New Developments in Modelling Soot Formation in Engines



Dr Markus Kraft

Sebastian Mosbach, Jon Etheridge, Abhijeet Raj, Richard West,  
Matt Celnik, Rob Patterson, Tim Totton, Markus Sander, Raphael  
Shirley, Neal Morgan, Amit Bhave, Andrew Smallbone

16 March 2009



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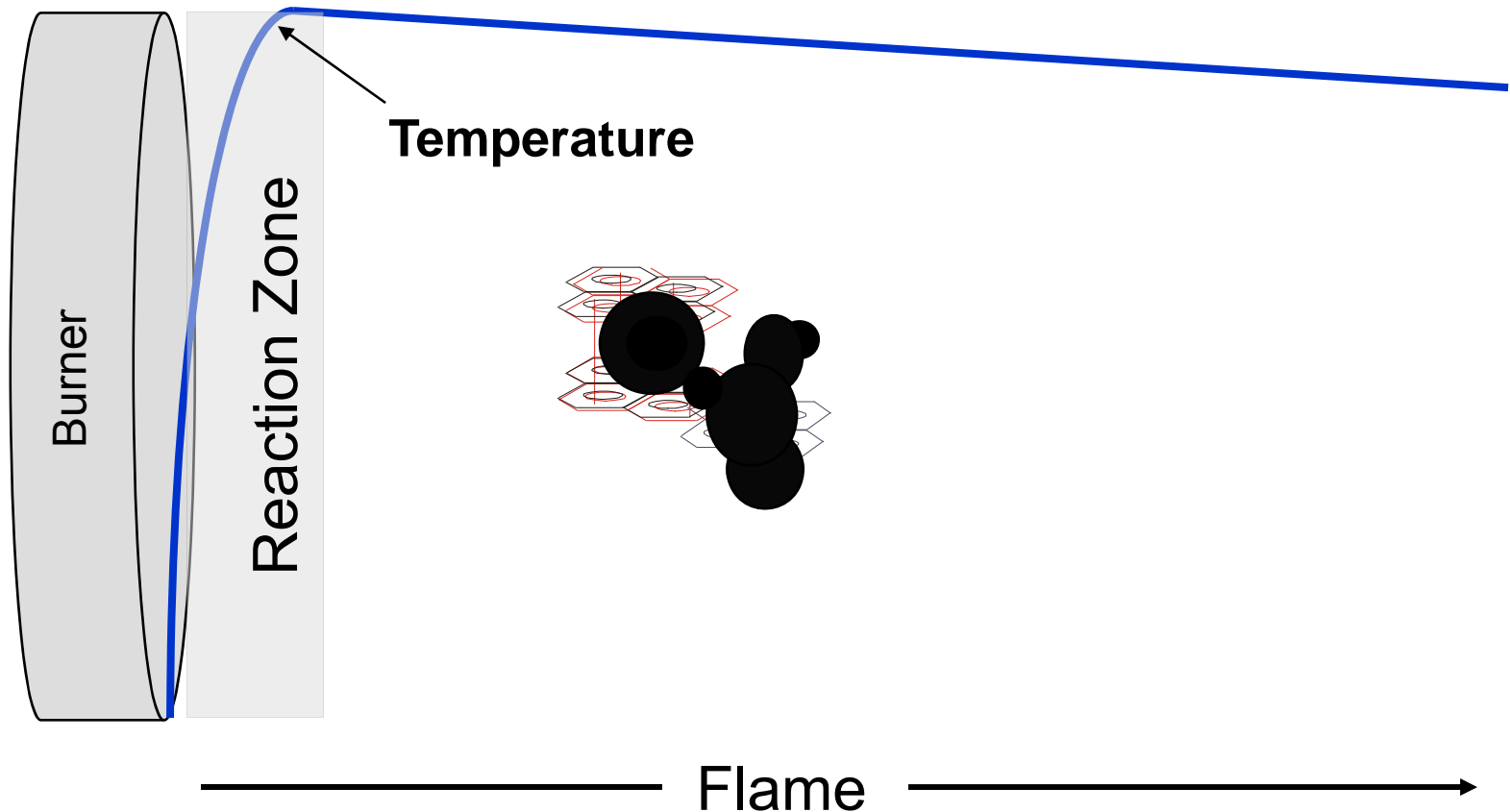
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CAMBRIDGE**

# Part 1:

- Quantum chemistry
- Kinetic Monte Carlo
- Basin hopping
- Population balance



# Soot Formation



Carbon Addition Reactions  
Oxidation by  $O_2$  and  $OH$ ,  
Aggregation

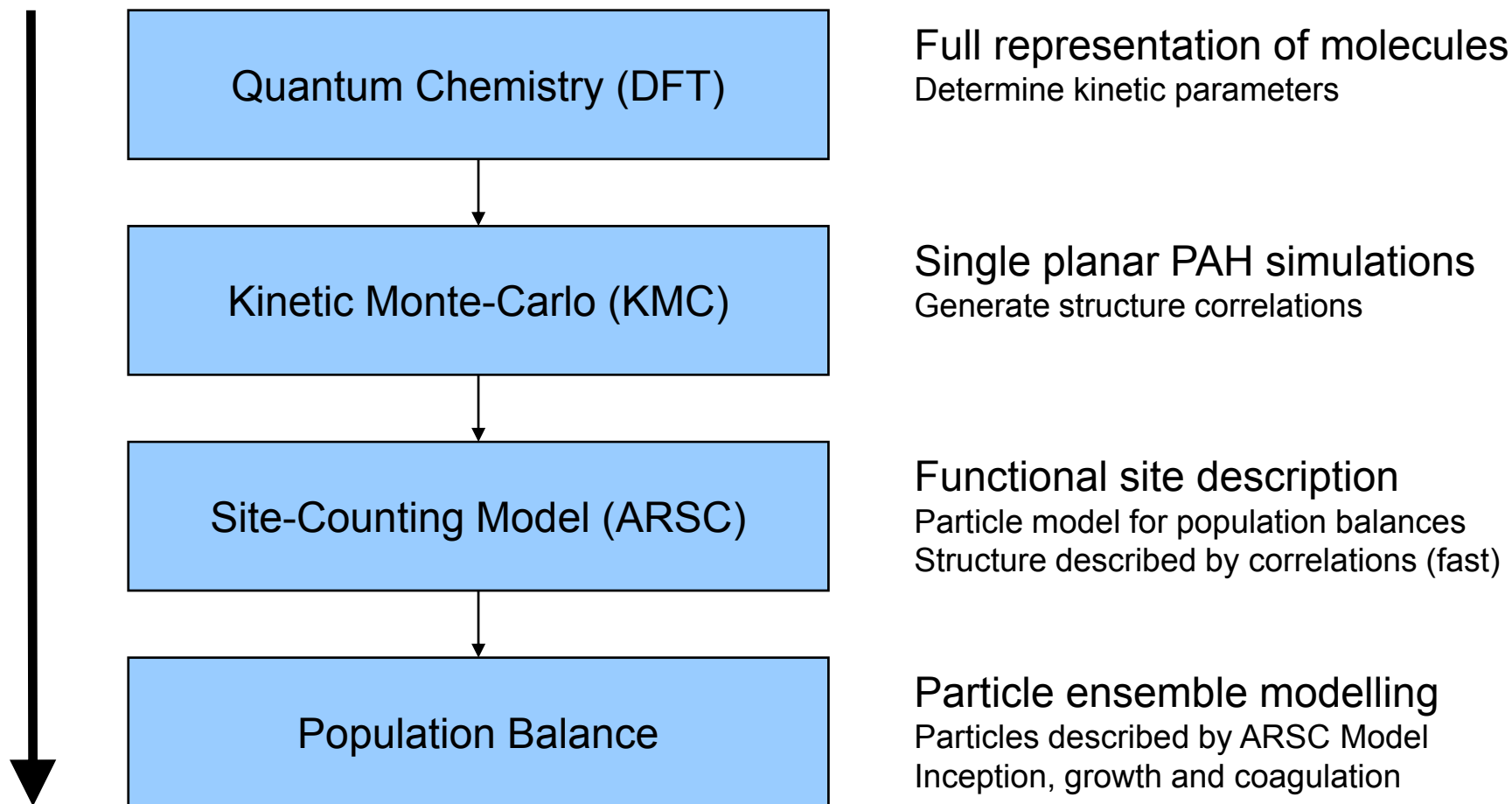


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# Soot model hierarchy

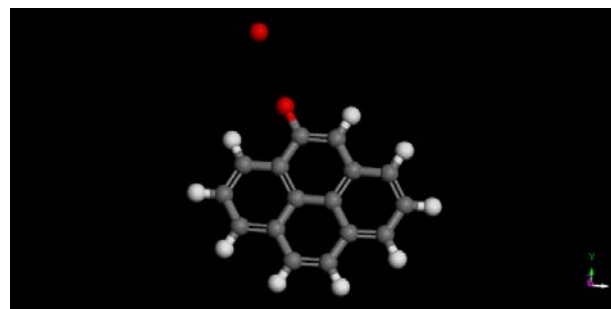
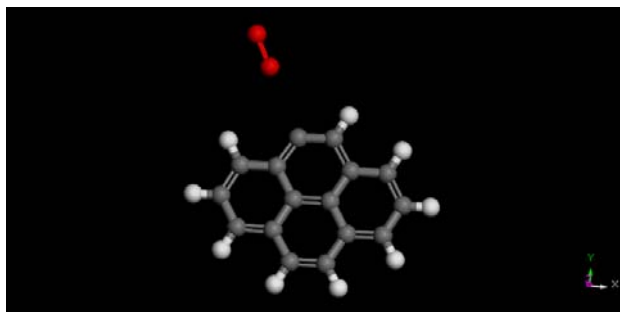


# Oxidation processes in PAHs

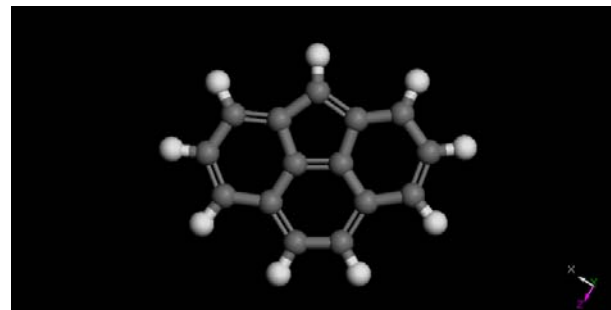
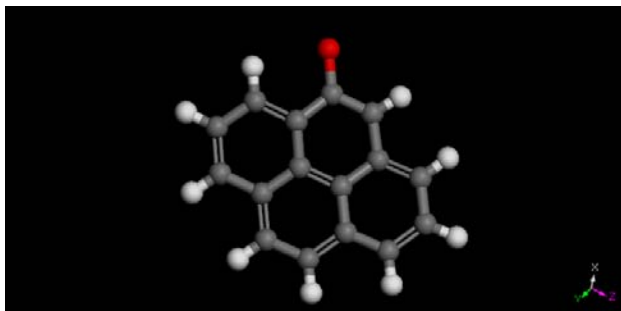
## Investigated reactions:

PAH: Polyaromatic hydrocarbons

Oxidation process:



Decomposition process:



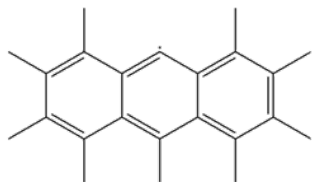
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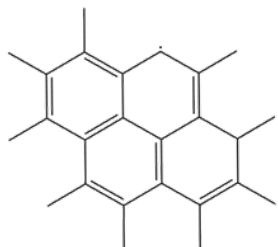
# Oxidation rates of different site types

Zigzag next to zigzag (zz)



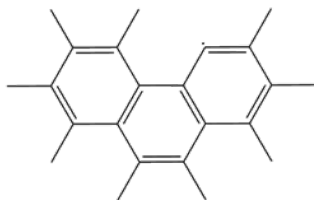
$E_{\text{act}}=156$  kJ/mole

Zigzag next to free edge (zf)

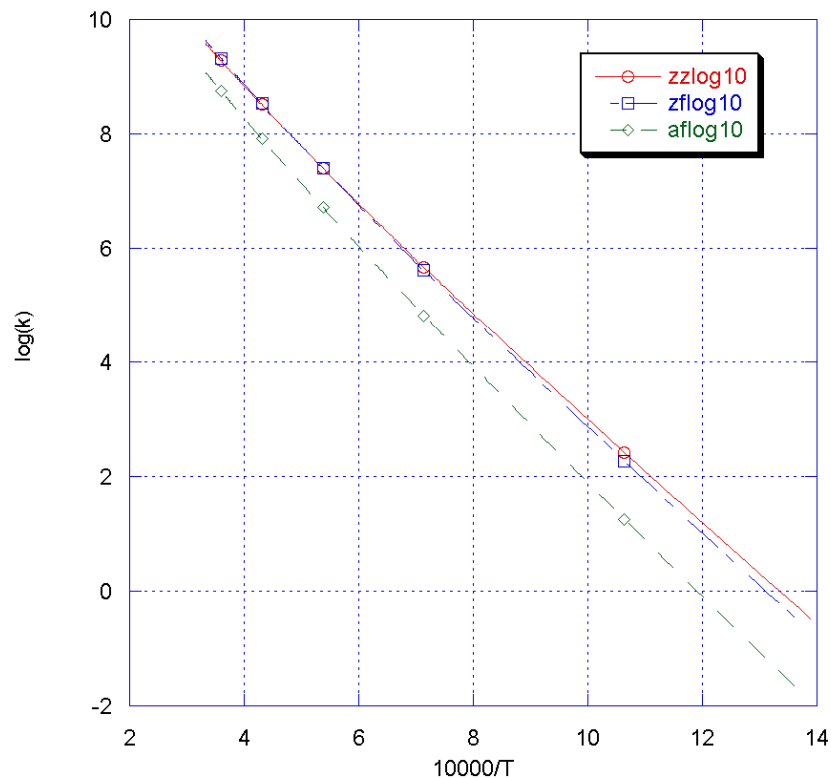


$E_{\text{act}}=161$  kJ/mole

Armchair next to free edge (af)



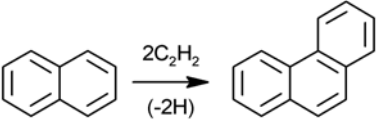
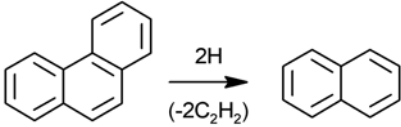
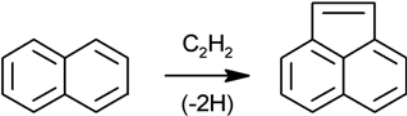
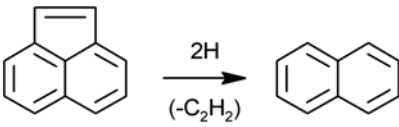
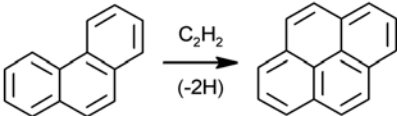
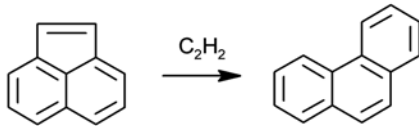
$E_{\text{act}}=173$  kJ/mole



Units:  $k$  in  $\text{cm}^3/(\text{mole}\cdot\text{s})$ ,  $T$  in K

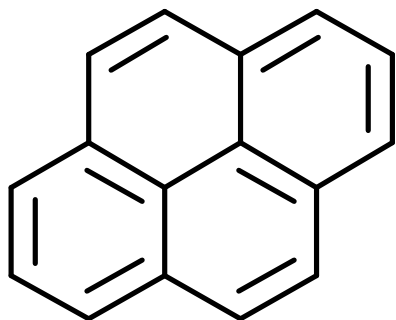


# PAH reactions (selection)

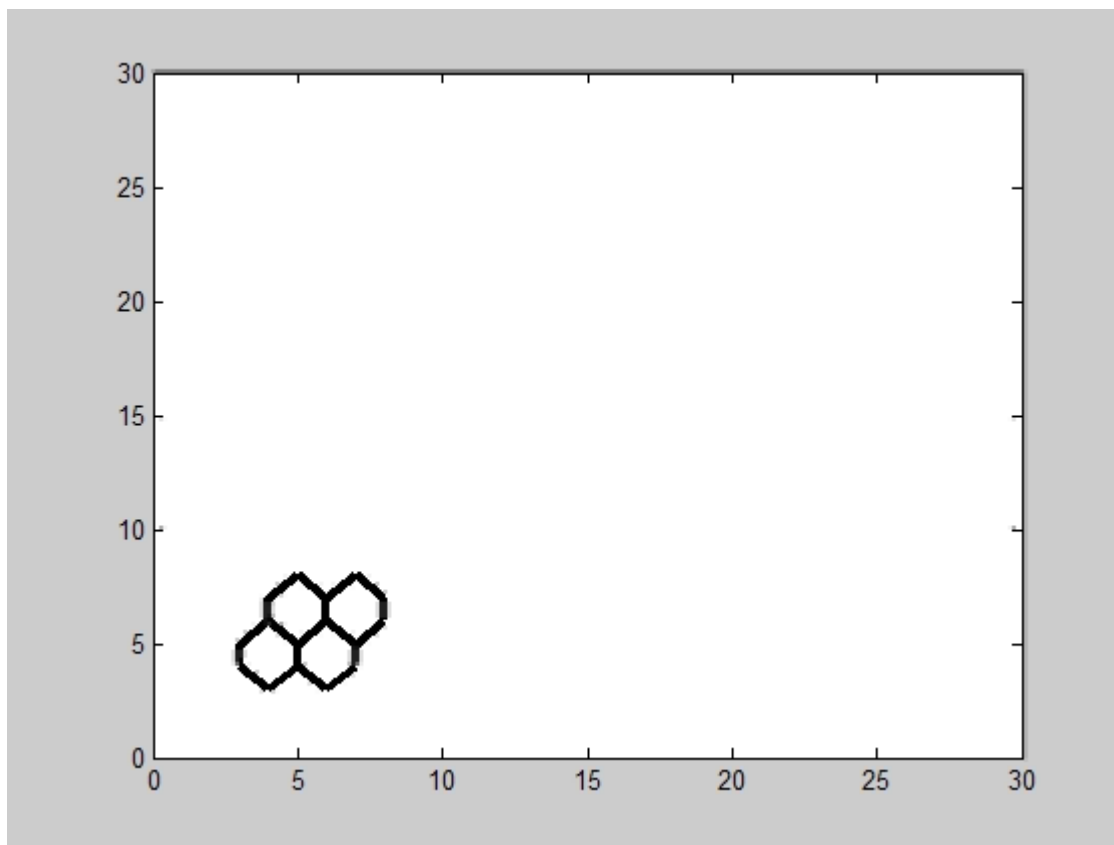
S1	Free-edge growth	
S2	Free-edge desorption	
S3	5-member ring addition	
S4	5-member ring desorption	
S5	Armchair growth	
S6	5- to 6-member ring	



# PAH growth simulation



Seed molecule:  
Pyrene

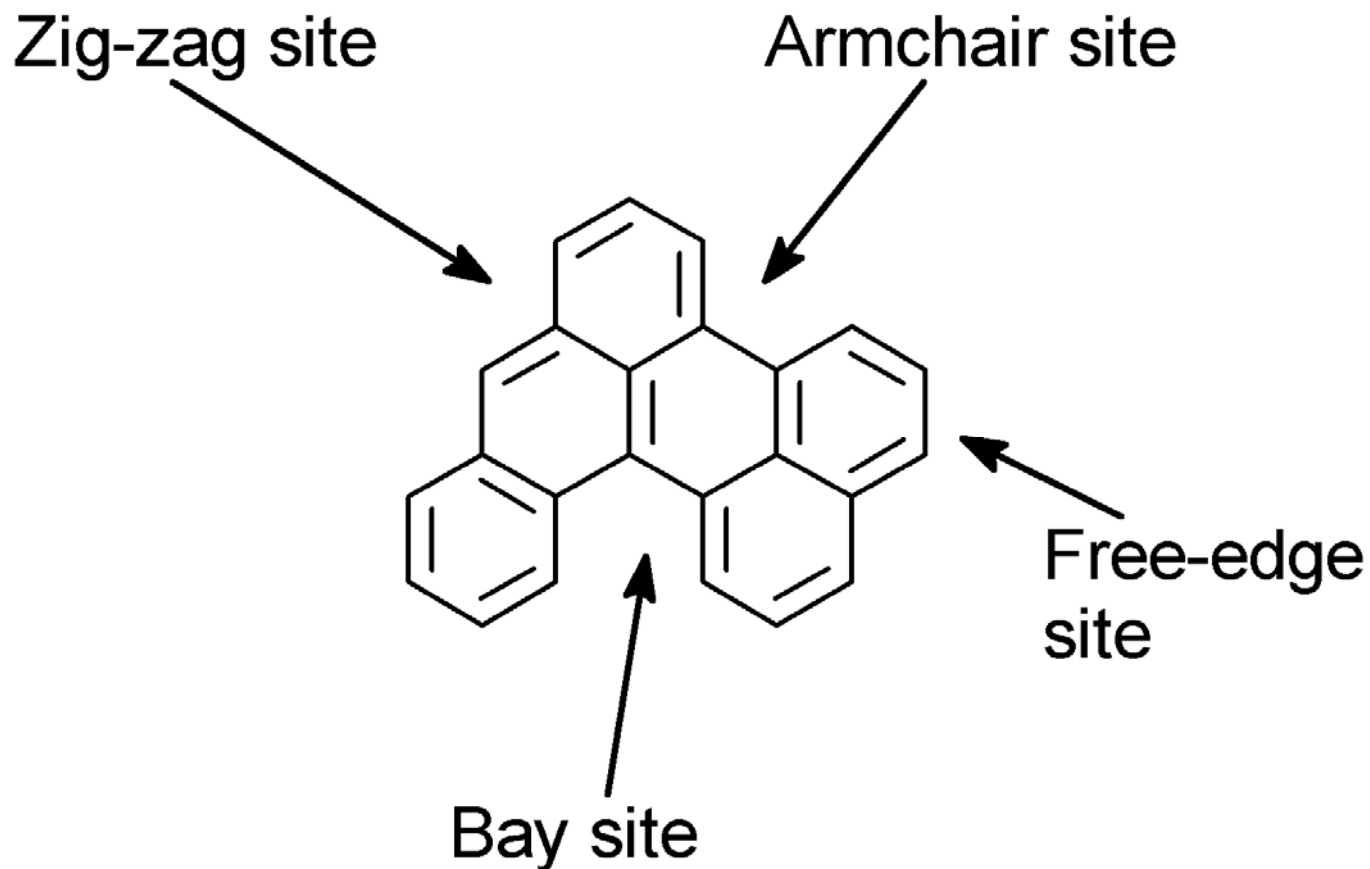


Growth of a PAH molecule



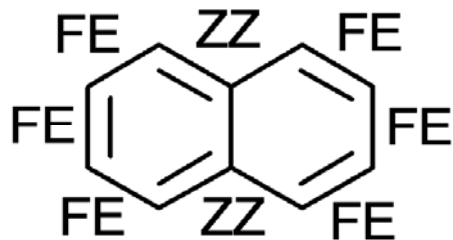


# Adding structural information

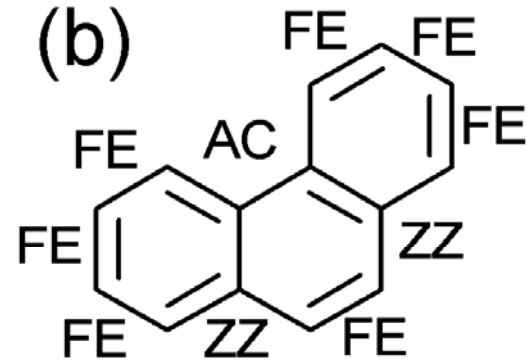


# S1: Free edge growth

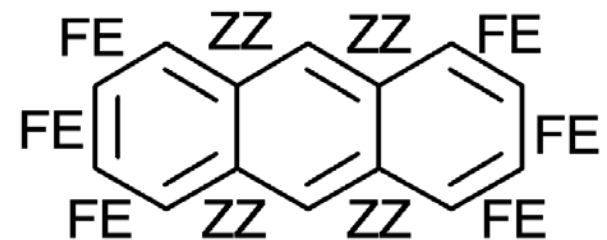
(a)



(b)



(c)



(a) 6 FEs, 2 ZZs

(b) 7 FEs, 2 ZZs, 1 AC

(c) 6 FEs, 4 ZZs

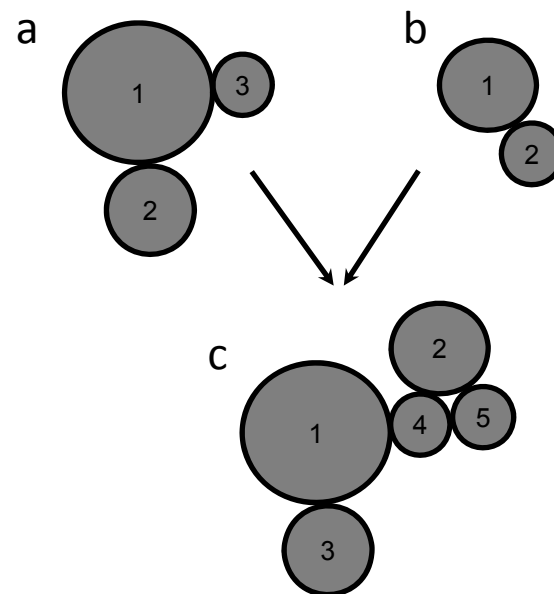
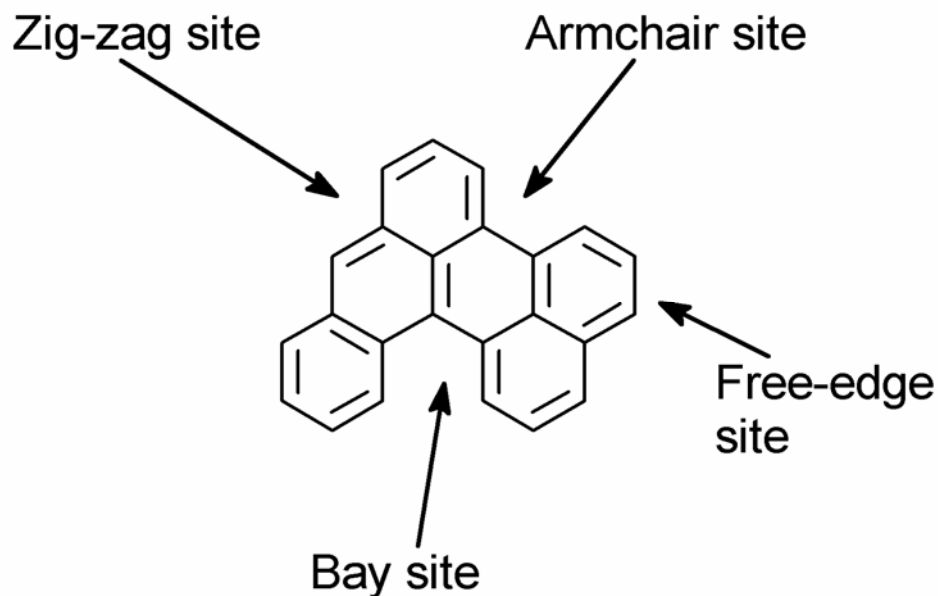


# Particle state space

Describe soot particles by  $9+N$  dimensional state space (ARS-SC-PP model):

$$E = (C, H, N_{FE}, N_{ZZ}, N_{AC}, N_{BY}, N_{R5}, S_a, N_{PAH}, PP_{(1-N)})$$

$PP$  = primary particle list





# Molecular soot structure

- Nanoparticles (3 nm – 2  $\mu\text{m}$ )
- ‘Primary particles’ occur due to dimerisation of PAH molecules
- What is the critical PAH size to form a dimer in flame conditions?
- How mobile are PAH molecules within a particle?
- How does morphology change and how does this affect the rates of sintering between primary particles?

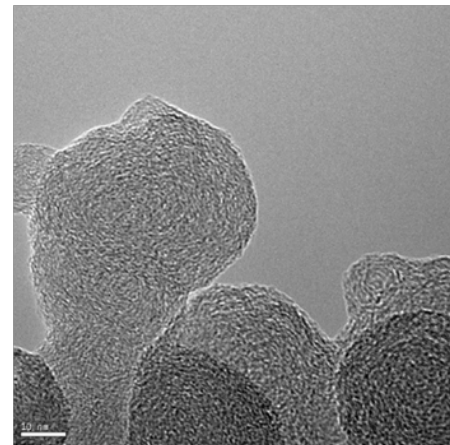
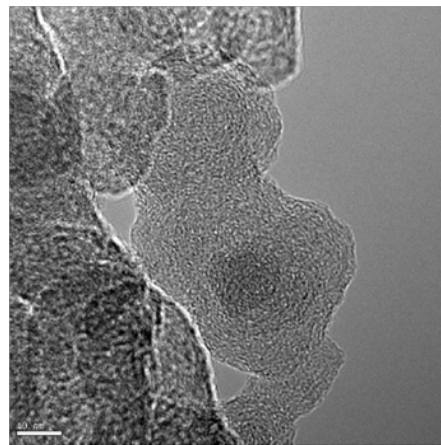
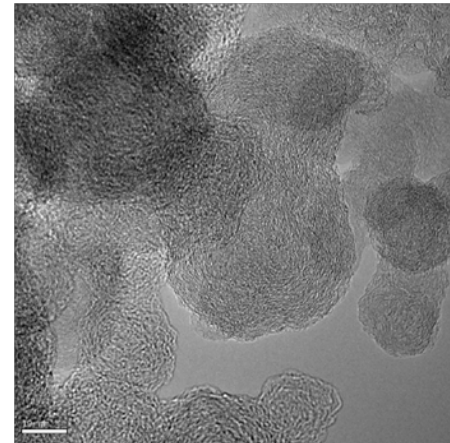
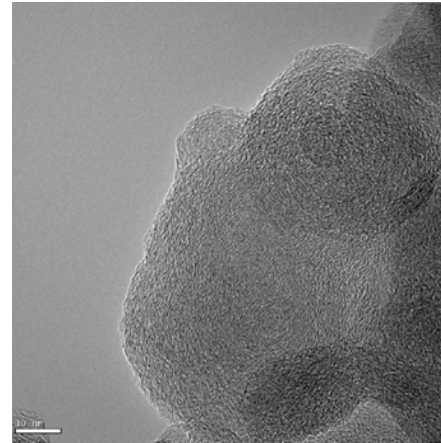




# TEM images of soot

- Some evidence for different soot structures based on different fuels
- Top: 'graphitic'
- Bottom: 'amorphous'

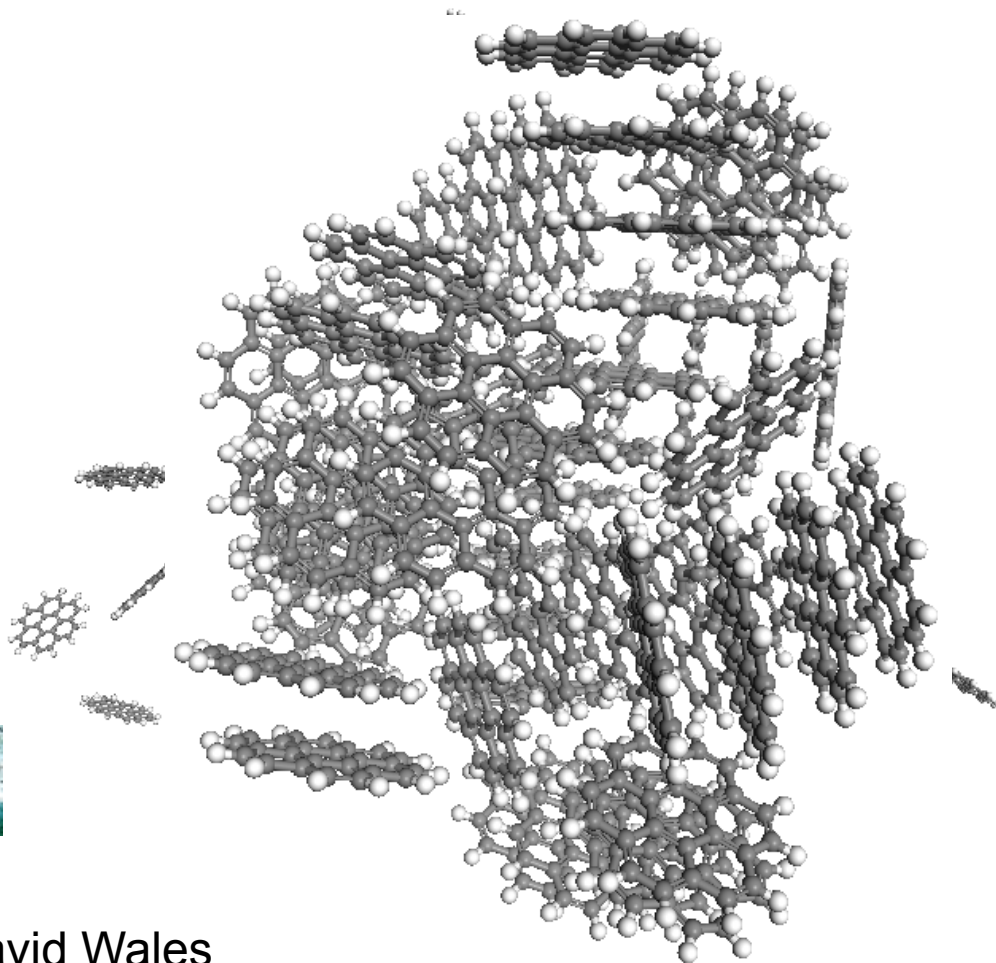
Pictures from: Uitz, Cracknell, Jansma and Makkee, "Impact of diesel fuel composition on soot oxidation Characteristics", SAE 2009-01-0286





# Investigating structure

- How do PAH molecule form clusters?
- Driven by intermolecular potentials
- How do these clusters grow?



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Alston Misquitta, Aron Cohen,  
Dwaipayan Chakrabarti, Mark Miller, David Wales



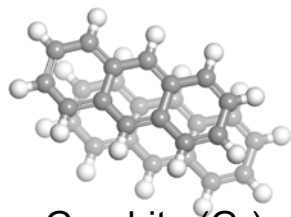
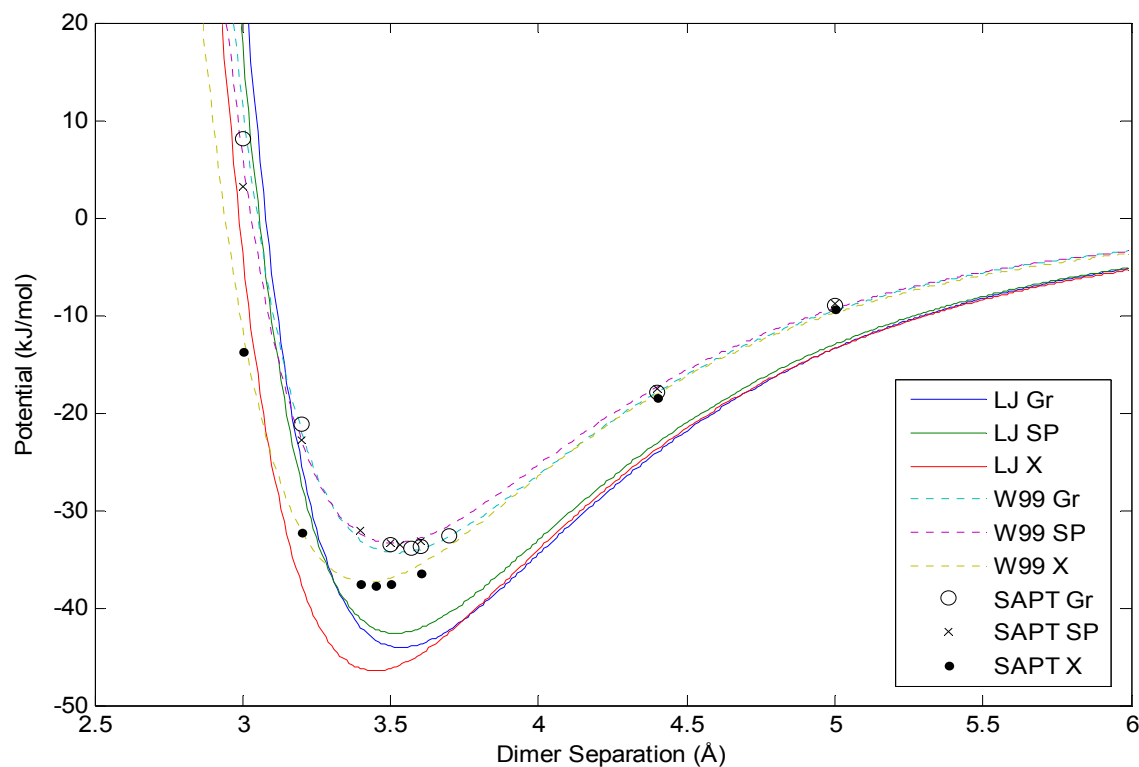
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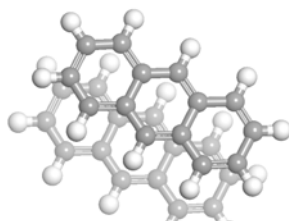


# Comparison of potentials

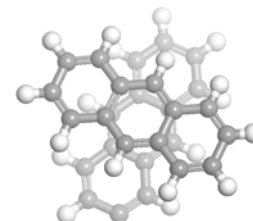
- Poor agreement between L-J potential and SAPT(DFT) results
- Good agreement with W99 potential



Graphite (Gr)



Slipped Parallel (SP)



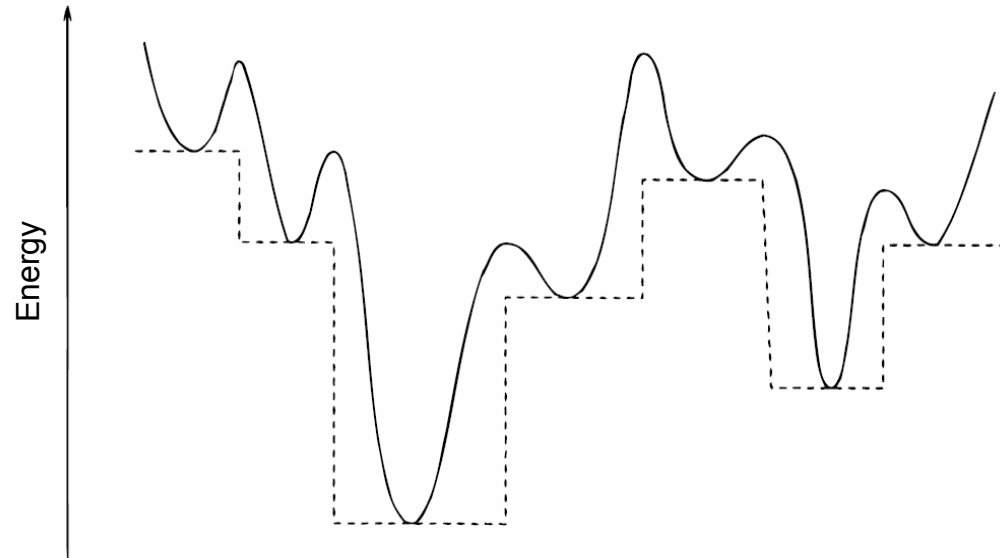
Crossed (X)





# basin hopping

- Finds stable molecular clusters by searching for minima
- Based on potential energy 'landscape'
- Uses Monte-Carlo criterion when 'jumping' between minima

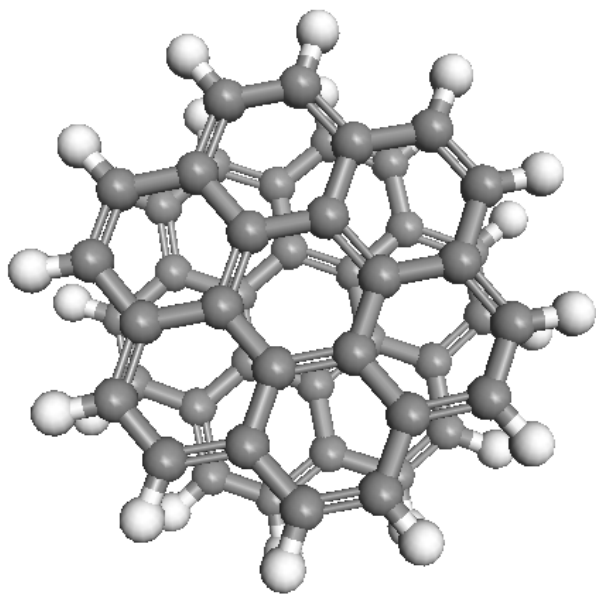






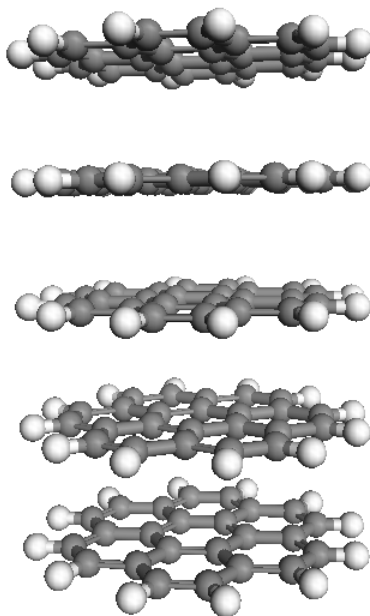
# global minimum clusters

2 Coronene molecules



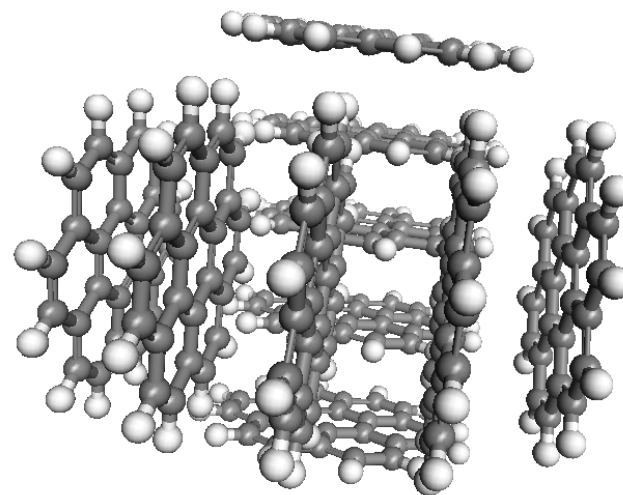
$$E = -94.90 \text{ kJ/mol}$$

5 Coronenes molecules



$$E = -394.35 \text{ kJ/mol}$$

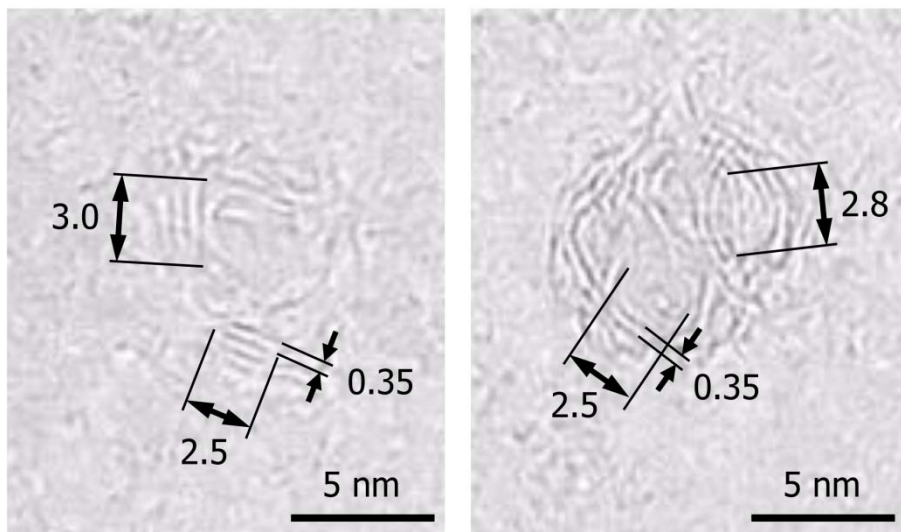
10 Coronene molecules



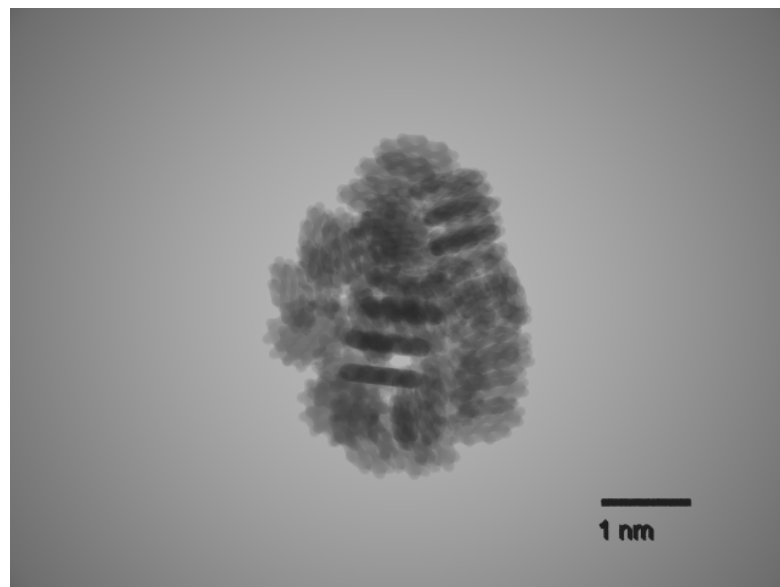
$$E = -926.42 \text{ kJ/mol}$$



# Experimental comparison



Experimental HR-TEM images of an aggregate sampled from a diesel engine. Indicated are length scales of structures within a primary particle (from Mosbach *et al.*, 2009, Combustion and Flame).



A TEM-style projection of a cluster of 50 coronene molecules



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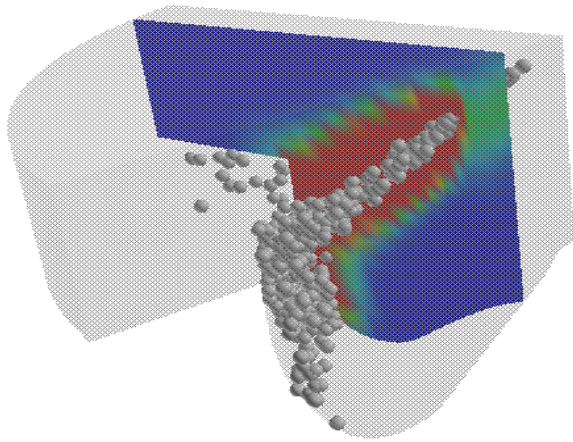


# Part 2:

- Soot model validation in SRM
- Converted SI – HC/CI engine
- SRM – SI
- Soot in DISI



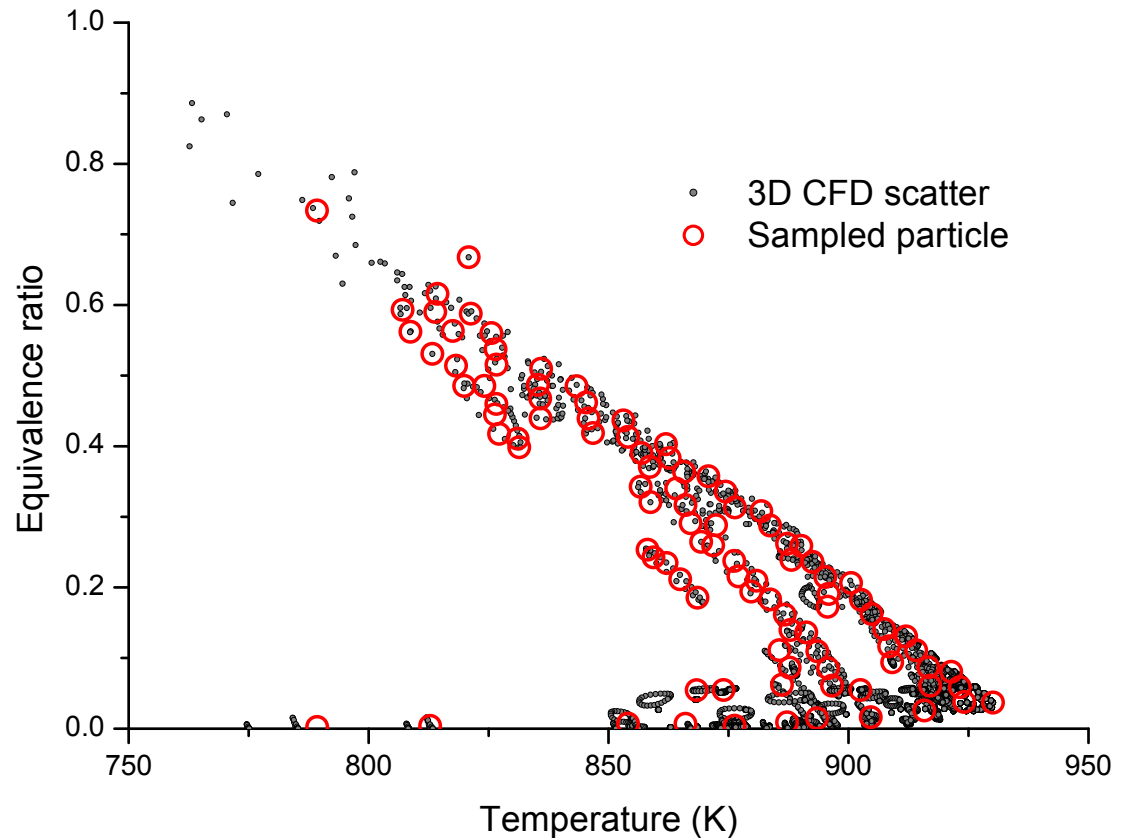
# Engine model: SRM



Diesel engine

SOI= -50  $\square$  ATDC

Spray Cone angle=100  $\square$

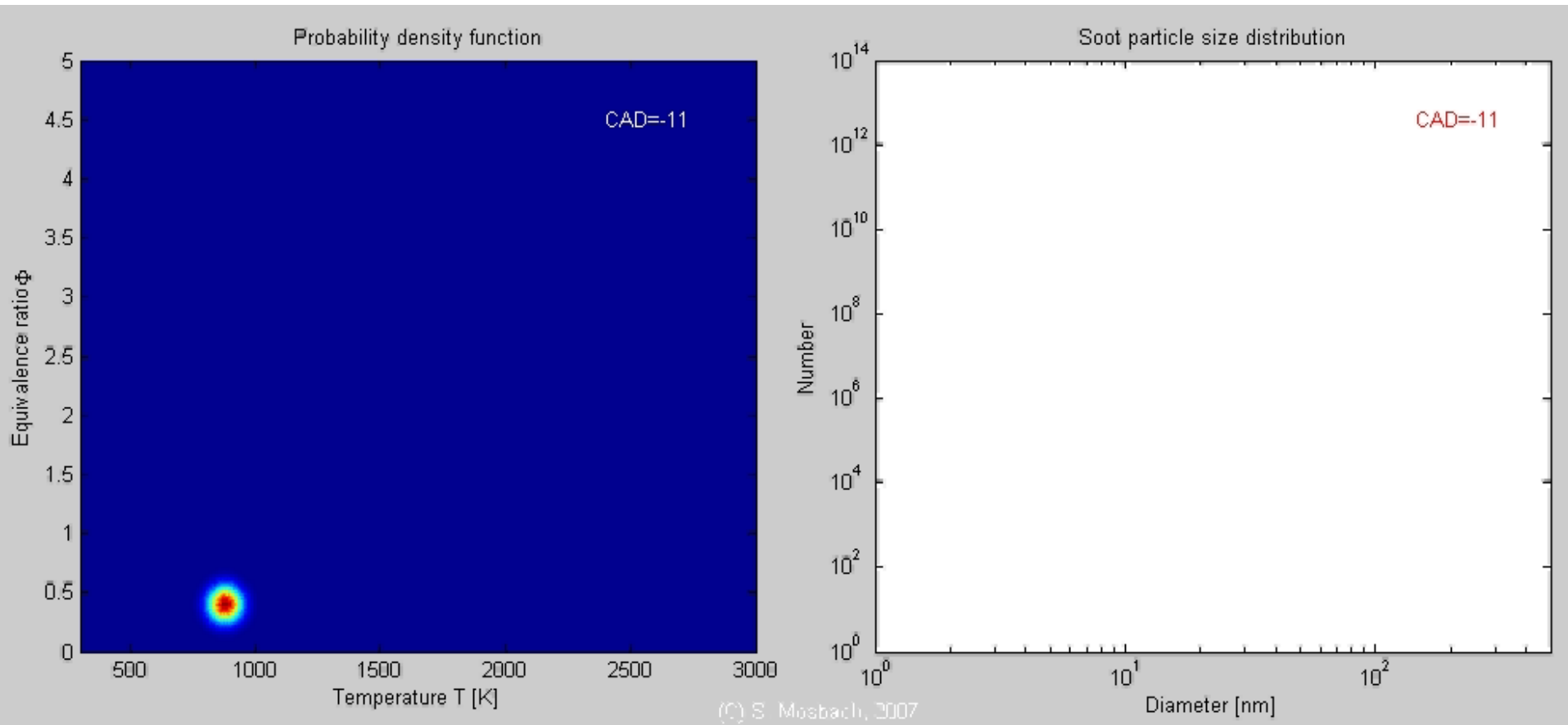


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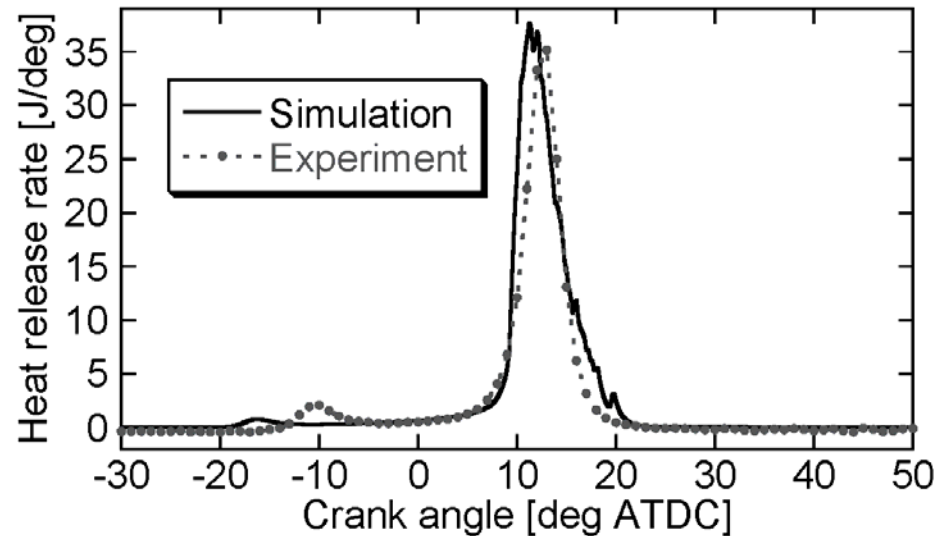
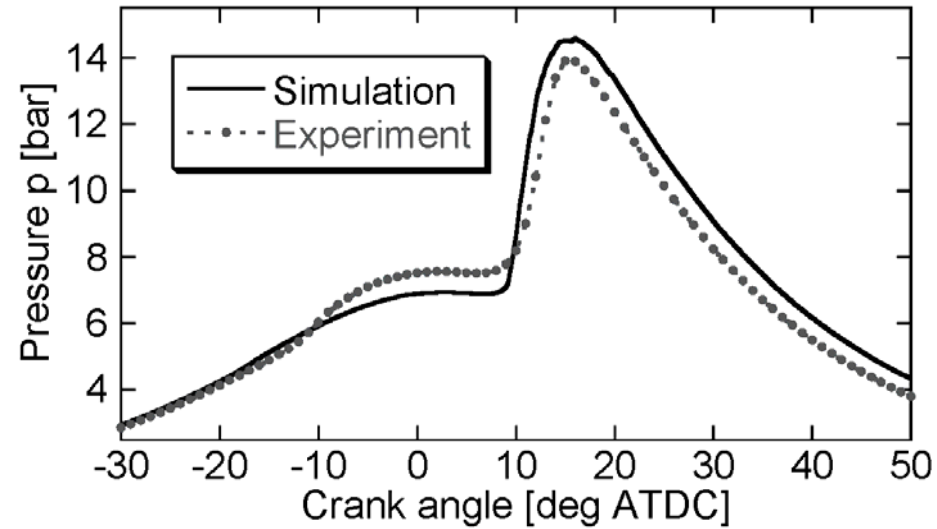
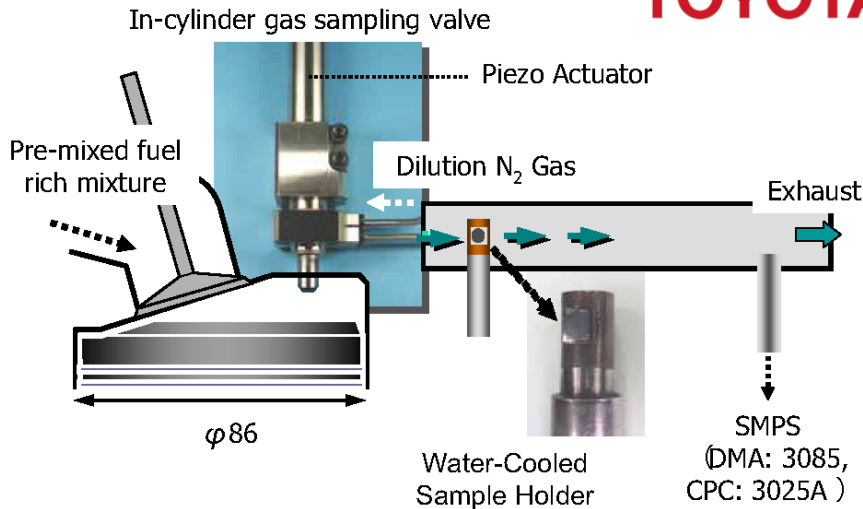


# Partially stratified HCCI engine



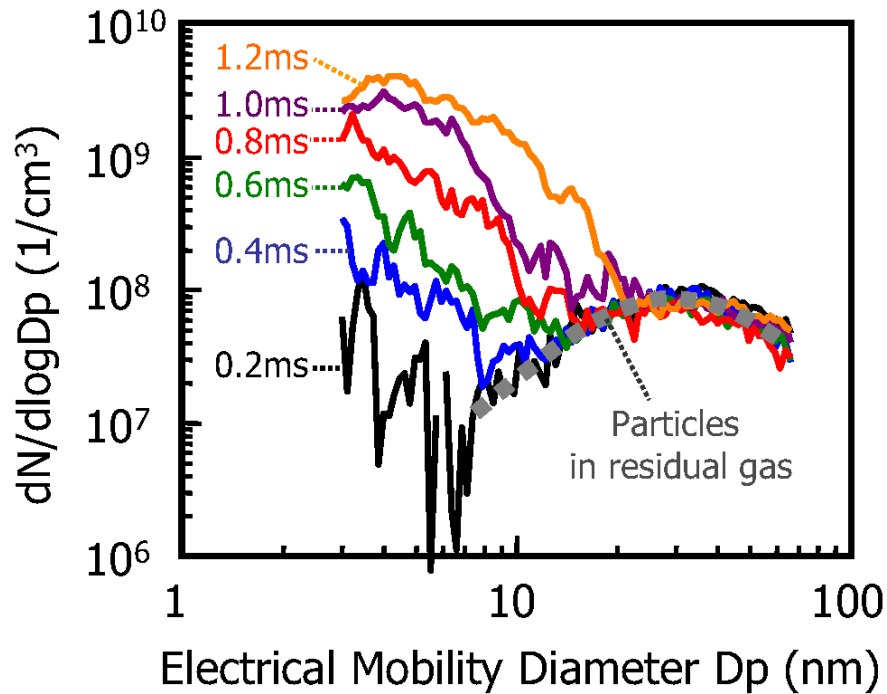
# Soot in engines!

- HCCI, n-heptane
- Compression ratio 12
- Equivalence ratio 1.93
- Throttled, 20% EGR

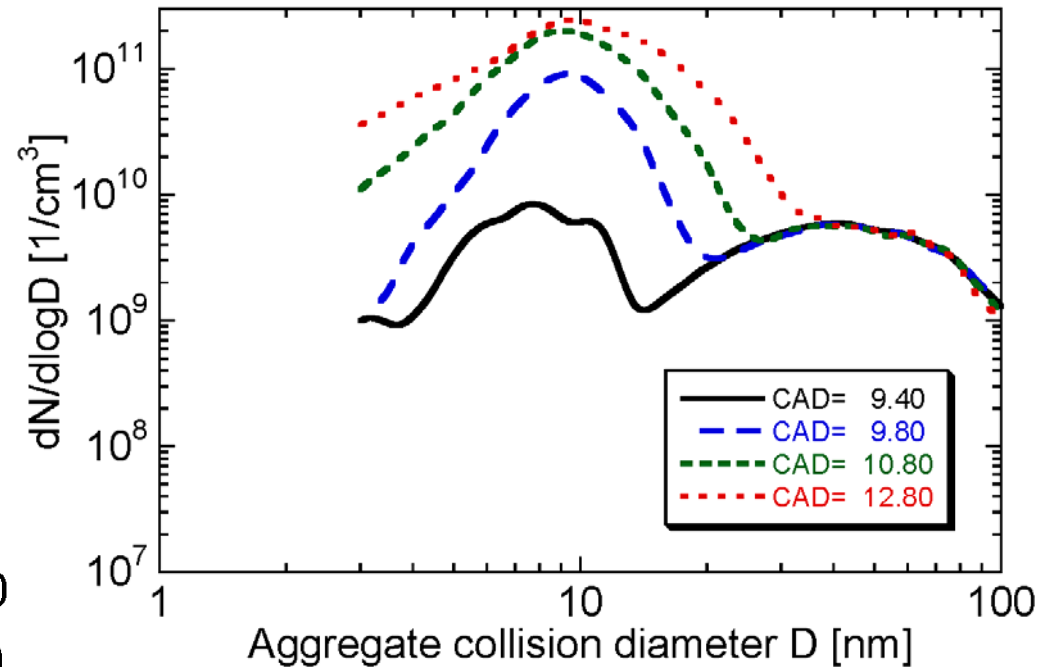


# Aggregate size distributions (I)

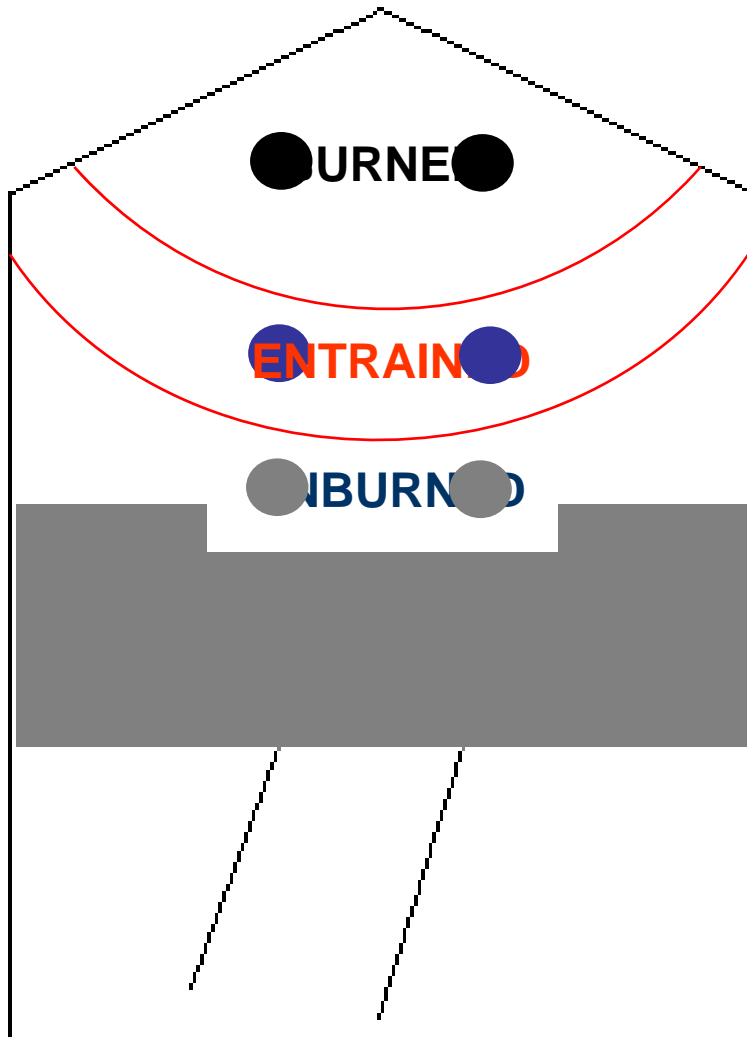
## Experiment



## Simulation



# SI Model

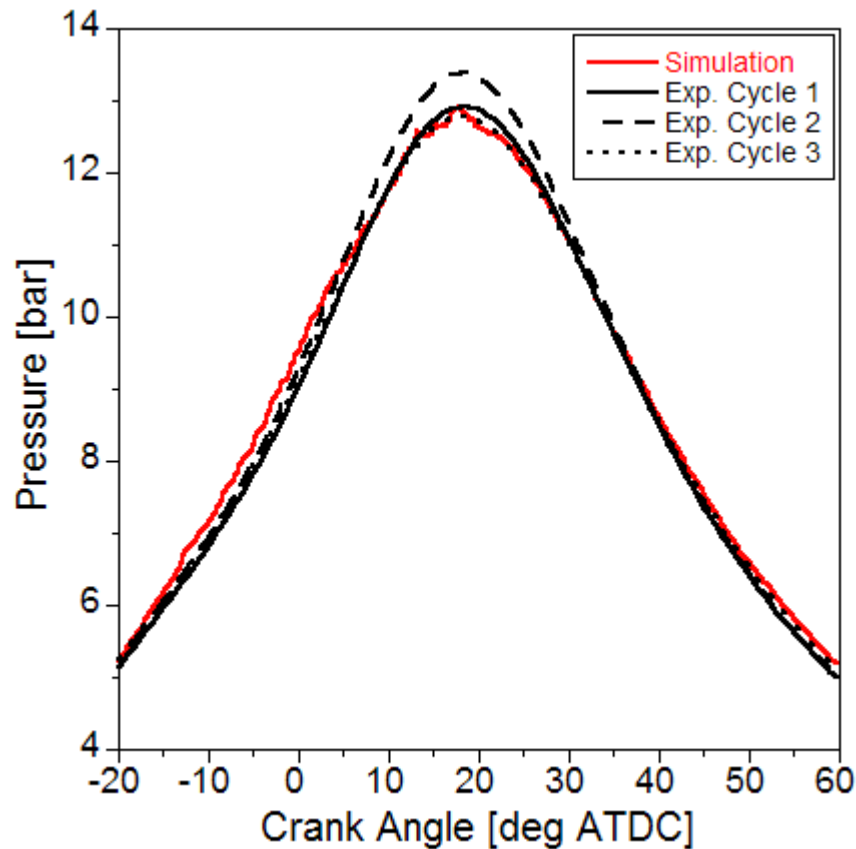


- The particles are filtered in the entrained zone and calculated with a detailed chemistry
- Mixing occurs within each zone but not between zones
- Equilibrium is established between zones and a new particle is added to the entrained zone.





# GDI SI Research Engine



In collaboration with Professor Richard Stone from Oxford University.

Fuel	iso-octane
Bore	89.0 mm
Stroke	90.3 mm
Con. rod length	148.9 mm
Disp. volume	562 cm <sup>3</sup>
CR	11.1
Speed	1500 RPM
Air/fuel equiv. ratio	1.0
EGR	15%

	Exp.	Sim.
uHC [ppm]	600	239
CO [ppm]	2700	2541
NO <sub>x</sub> [ppm]	-	886



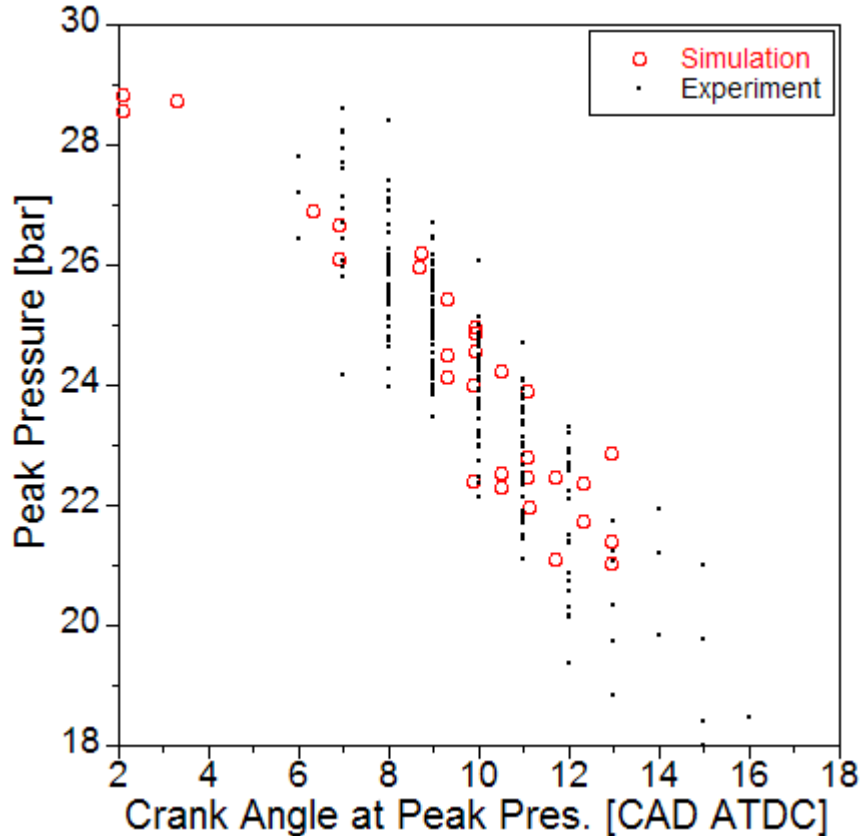
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# Multi-cycle SI Simulation



- Model coupled to GT-Power for multi-cycle simulation.
- 40 simulated and 200 experimental cycles.
- NO<sub>x</sub> emissions:
  - 496 ppm simulation
  - 528 ppm experiment



# DISI Engine



- Late injection produces stratified mixture.
- Fuel rich regions close to spark gap.

Image from [www.engineforall.com](http://www.engineforall.com)

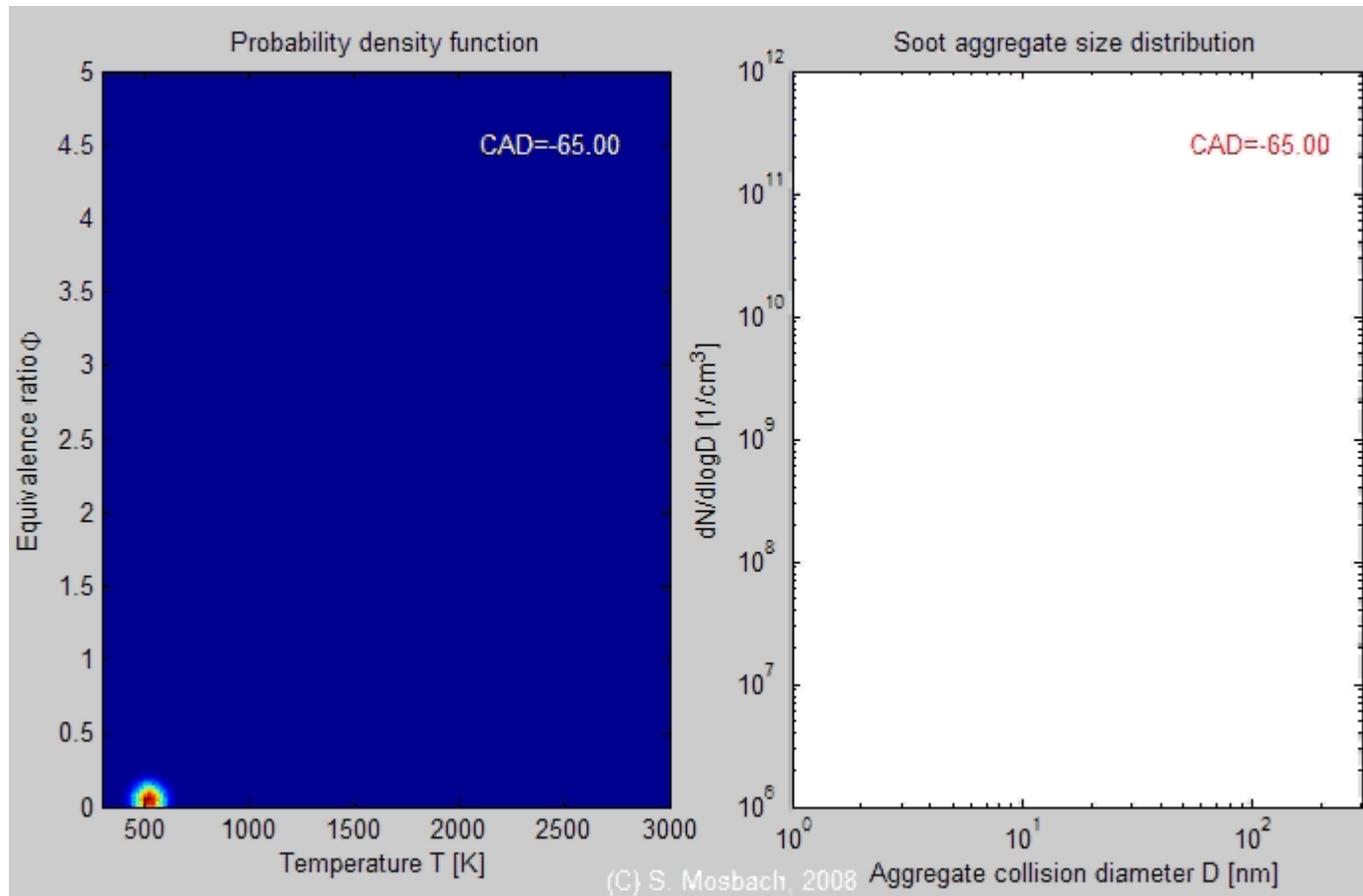


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# Soot in DISI Engine



$$\lambda = 1.0$$

EOI = -50 CAD ATDC

Spark = -30 CAD ATDC



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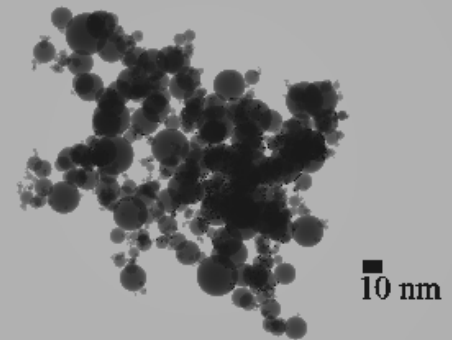
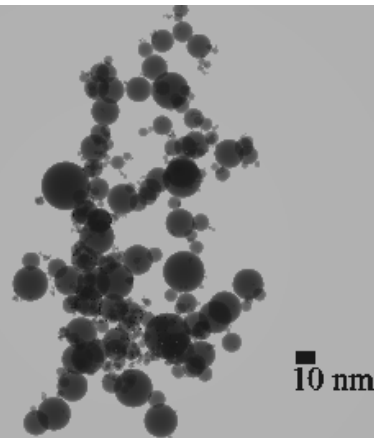
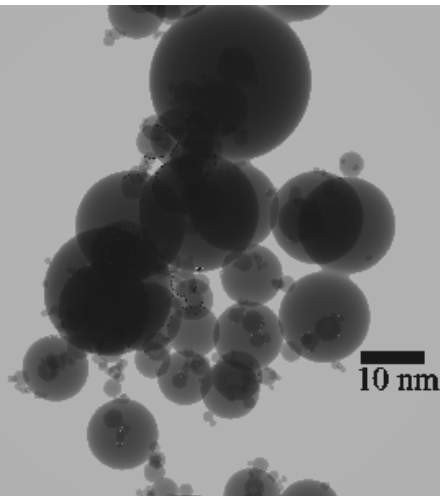


# Soot in DISI Engine

2.6 CAD ATDC

12.6 CAD ATDC

32.6 CAD ATDC



CAD [deg ATDC]	2.6	12.6	32.6
No. Primaries	492	1315	2083
Coll. Diam [nm]	70	108	137



# Thank you!

## Please visit our website:



The screenshot shows the homepage of the CoMo Group at the University of Cambridge. The header includes the CoMo Group logo (two dice) and the text 'CoMo GROUP'. To the right is the University of Cambridge crest and 'UNIVERSITY OF CAMBRIDGE DEPARTMENT OF CHEMICAL ENGINEERING'. A navigation bar contains links: Home, People, Research, Resources, Preprints, Publications, Conferences, Seminars, Login. Below this is a sub-navigation bar with 'Welcome', 'Vacancies', and 'Contact'. The main content area starts with 'Welcome!' followed by a portrait of Markus Kraft. To the right of the portrait is a welcome message: 'Welcome to the website of the Computational Modelling Group! We develop and apply modern numerical methods to problems arising in Chemical Engineering. The overall aim is to shorten the development period from research bench to the industrial production stage by providing insight into the underlying physics and supporting the scale-up of processes to industrial level.' Below this is a paragraph: 'The group currently consists of 20 members from various backgrounds. We are keen to collaborate with people from both within industry and academia, so please [get in touch](#) if you think you have common interests.' The final paragraph states: 'The group's [research](#) divides naturally into two inter-related branches. The first of these is research into mathematical [methods](#), which consists of the development of stochastic particle methods, computational fluid dynamics and quantum chemistry. The other branch consists of research into [applications](#), using the methods we have developed in addition to well established techniques. The main application areas are reactive flow, combustion, engine modelling, extraction, nano particle synthesis and dynamics. This research is [sponsored](#) on various levels by the UK, EU, and industry.'



Markus Kraft - Head of the CoMo Group

## <http://como.cheng.cam.ac.uk>



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